Synthesis of *trans*-2,6-Dialkylpiperidines by 1,3-Cycloaddition of Alkenes to 2-Alkyl-2,3,4,5-tetrahydropyridine Oxides

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A convenient route to *trans*-2,6-dialkylpiperidines by cycloaddition of alkenes to 2-alkyl-2,3,4,5-tetrahydropyridine oxides followed by reductive cleavage of the resulting isoxazolidine is illustrated by a synthesis of the fire ant-venom alkaloid, solenopsin.

reports¹ on the synthesis of trans-2,6dialkylpiperidines from tetrahydropyridine oxide by cycloaddition reactions prompt us to record some of our own results² in this area. We too had conceived the possibility that cycloaddition of an alkene to a 2-alkyl-2,3,4,5-tetrahydropyridine 1-oxide would take place preferentially by orthogonal approach of the alkene to the nitrone in a conformation in which the 2-alkyl substituent was equatorial, to give an isoxazolidine which would furnish a trans-2,6-dialkylpiperidine by reductive cleavage of the N-O bond. We have shown the validity of this supposition by a short stereocontrolled synthesis of solenopsin (1), one of the constituents of the venom of the fire ant Solenopsis saevissima.³ Related results were reported by Gossinger⁴ during the course of our work and he used the reaction in a neat synthesis of the alkaloid porantherilidine.

In our approach to solenopsin, 2-methyl-2,3,4,5-tetrahydropyridine 1-oxide, obtained from 2-methyl-1-hydroxypiperidine with mercuric oxide, was treated with undec-1-ene in chloroform at 50 °C to give the isoxazolidine (2) in 47% yield after chromatography. Thin layer and gas-liquid chromatography and the ¹³C n.m.r. spectrum clearly indicated the formation of only one stereoisomer in this reaction, shown to be the *trans* compound (2) by conversion into solenopsin. The stereochemistry at C-5 in (2) is assigned by analogy. ⁵ Reductive cleavage of the isoxazolidine ring with

zinc and acetic acid afforded the piperidine derivative (3), again as a single isomer. The chemical shifts of the α and α' carbon atoms (δ 46.0 and 47.7) in the ¹³C n.m.r. spectrum of (3) closely resembled those reported for solenopsin (δ 45.9 and 50.9) rather than those of isosolenopsin with *cis* alkyl substituents (δ 52.6 and 52.7).6 Conversion of (3) into solenopsin was effected by reduction of the corresponding phenoxythioxocarbonate with tri-n-butylstannane⁷ or, better, by desulphurisation of the thio-acetal of the derived ketone (4)

with Raney nickel and acid hydrolysis of the carbamate. None of the *cis* isomer was detected by g.l.c.

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